

SAULT COLLEGE OF APPLIED ARTS & TECHNOLOGY
SAULT STE. MARIE, ONTARIO

COURSE OUTLINE

Course Title: Metallurgy
Code No,: MET 112
Program Machine Shop
Semester: Two
Date: 1986 05 14
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New XX Revision

APPROVED

D.P. Crozetta
Chairperson

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Date

Metallurgy

Met 112

Course Name

Course Number

PHILOSOPHY/GOALS ;

When the student has successfully completed this course of study, he/she should have a reasonable understanding of the material presented. The intention (therefore) is to provide students with sufficient background to assist others in the solution of work related (metallurgical) problems.

METHODS OF ASSESSMENT (GRADING METHOD):

3 Theory Tests	70%
1 Lab Report	20%
Attendance/Attitude	10%

(with **NO** incompletes)

TEXTBOOK(S) :

"Technology of Machine Tools" (ch 12)
2nd edition, McGraw-Hill, Ryerson

OBJECTIVES :

The basic objective is to develop within the student an understanding of the concepts and procedures involved with this course of study as well as an ability to use them in the solution of problems. Theory tests and lab exercises are designed with this in mind.

The basic level of competency demanded is an over-all course average of 60% with no incompletes.

TOPIC NO.	PERIODS	TOPIC DESCRIPTION	REFERENCE
	1 1/2	INTRODUCTION & ORIENTATION	
		- course topics	Handout
		- general objectives	
		- methods of evaluation	
		- grading system	
		- teaching methods	
		- policy regarding	
		a) attendance	
		b) attitude	
		c) due dates	
		d) re-writes	
		e) testing	
		PRODUCTION OF IRON & STEEL	
		- iron ore minerals, chemical formula and gangue materials	Text Ch 17
		- iron production via blast furnace reduction	P397-401
		- types of steelmaking furnaces	
		- general types of commercial metals and chemical analysis	
		THEORY TEST # 1 FOR TOPICS 1 & 2	
	5	THE IRON; IRON-CARBIDE SYSTEM	
		- general understanding of the iron;	Text
16-L		iron-carbide system for steels	CH 17
		- changes in steels as they are heated	P409-413
		- temperature ranges for heat treatment	Handout
		- requirement to harden steels	
		- formation and hardness of martensite	
	2	THEORY TEST # 2 FOR TOPIC # 3	

TOPIC NO.	PERIODS	TOPIC DESCRIPTION	REFERENCE
		SURFACE TREATMENTS	
		- purpose and methods of carburizing	Text
		effects of carburizing on steels	CH 17
		purpose of flame and induction hardening	P413- 416
		effects of flame and induction hardening on steels	
		NON-FERROUS METALS	
		- general understanding of the aluminum-copper system	Handout
		- requirements to harden aluminum-copper alloys	
		- general steps of hardening process	
		THEORY TEST # 3 FOR TOPICS 4 & 5	
		LAB EXPERIMENTS	
		ROCKWELL HARDNESS & MICROSTRUCTURE(S)	
		- prepare and test samples for hardness	Handout
		- recognize hardness of samples with respect to carbon content and initial condition of sample	
		- prepare and observe samples for initial microstructure	
		- recognize microstructure with respect to initial condition of sample	
		- develop an understanding of the term NORMALIZE	
		- 6 SAMPLES REQUIRED/GROUP	

TOPIC NO.	PERIODS	TOPIC DESCRIPTION	REFERENCE
2	4	<p>ANNEALING & HARDENING</p> <ul style="list-style-type: none"> - heat treat samples for the purpose of changing their hardness and microstructure - prepare and test samples for hardness - recognize changes in hardness with respect to annealing and hardening - prepare and test samples for microstructure - recognize changes in microstructure with respect to annealing and hardening - develop an understanding of the terms ANNEAL/HARDEN - 6 SAMPLES REQUIRED FROM # 1 	Handout
3	4	<p>TEMPERING</p> <ul style="list-style-type: none"> - heat treat samples for the purpose of changing their "hardened" microstructure - prepare and test samples for hardness - recognize changes in hardness with respect to tempering - develop an understanding of the term TEMPER - 3 SAMPLES REQUIRED/GROUP FROM # 2 	Handout
4	4	<p>HEAT TREAT REVIEW</p> <ul style="list-style-type: none"> - review and discuss lab data - discuss changes with respect to iron-carbide system - discuss the terms: NORMALIZE, ANNEAL, HARDEN, TEMPER - complete lab reports 	Handout

NOTE: Course objectives and lab topics are subject to change with advance notice.

SPECIFIC OBJECTIVES

for

METALLURGY - 112

INTRODUCTION AND ORIENTATION - 1 1/2 HRS.

The student should be given an opportunity to:

- a) Identify and list the topics covered in this course outline.
- b) Identify and list the general objectives of this course outline,
- c) Identify and list the various methods of evaluation used in this course outline.
- d) Identify the grading system used in this course outline with respect to A, B, C, R, I, X.
- e) Identify the policy of this course with respect to:
 - i) attendance
 - ii) attitude
 - iii) due dates
 - iv) re-writes
 - v) testing policies
- f) Identify and list the various teaching methods used in this course outline.

PRODUCTION OF IRON AND STEEL - 2 HRS.

The student should be given the opportunity to:

- a) Name 4 iron ore minerals found in nature.
- b) Write the chemical formula that represents each of the iron ore minerals.
- c) List the various impurities and gangue materials found in iron ores.
- d) Name the furnace used to produce pig iron.
- e) Define the term "**reduction**" with respect to the blast furnace operation.
- f) List 3 major steelmaking furnaces in use today.
- g) List the general types of cast irons, cast steels and rolled steel sections in use today.
- h) Give the approximate carbon content, significant alloys and minor constituents found in cast irons, cast steels and rolled steel sections.

THE IRON; IRON-CARBIDE SYSTEM

The student should be given the opportunity to:

- a) Develop a general understanding of the iron; iron-carbide system for steels with respect to:
 - i) lower critical temperature
 - ii) upper critical temperature
 - iii) eutectoid point and composition
 - iv) existing equilibrium structures1 1/2 HRS
- b) Explain the changes in eutectoid, hypoeutectoid and hypereutectoid steels when they are heated from room temperature to above the upper critical temperature, 2 HRS,
- c) Identify and select the proper temperature ranges for the following heat treating operations:
 - anneal
 - normalize
 - harden
 - temper1/2 HR.
- d) List the three requirements necessary to successfully harden steels. 1/2 HR.
- e) Explain the formation of martensite. 1/2 HR.
- f) State the theory that explains why martensite has such a high hardness.

SURFACE TREATMENTS - 2 HRS.

The student should be given the opportunity to:

- a) State the purpose for which carburizing operations are carried out.
- b) State the 3 main carburizing processes.
- c) State the initial carbon content of steels used in carburizing operations.
- d) Describe the effects of the carburizing process on:
 - i) The "**final**" carbon content of the steels*
 - ii) The "**final**" microstructure and hardness of the steels.
- e) State the purpose for which flame hardening and induction hardening are carried out.
- f) State the initial carbon content of steels used in the flame and induction hardening processes.
- g) Describe the effects of flame and induction hardening processes on:
 - i) The "**final**" carbon content of the steels,
 - ii) The "**final**" microstructure and hardness of the steels.

5) NON-FERROUS METALS - 1 HR.

The student should be given the opportunity to:

- a) Develop a general understanding of the aluminum-copper system with respect to:
 - i) The aluminum-rich end.
 - ii) Temperature zone for solution treating,
 - iii) The temperature(s) for aging and artificial aging.
- b) List the three requirements necessary to harden aluminum-copper alloys.
- c) Describe or explain the following terms:
 - i) solution treat
 - ii) aging
 - iii) artificial aging

NOTE: SUBJECT TO CHANGE.